Cycloadditions with Tautomeric Systems: Conversion of Oxazol-4(5H)-ones into Substituted Furans

By K. T. Potts* and J. Marshall

(Department of Chemistry, Rensselaer Polytechnic Institute, Troy, New York 12181)

Summary 2-Phenyloxazol-4(5H)-one undergoes cyclo-addition with acetylenic dipolarophiles to give 3,4-disubstituted 2-phenylfurans in moderate yields.

The mesoionic oxazolium-5-oxides are amongst the most reactive mesoionic compounds in 1,3-dipolar cycloaddition reactions; this property is shared to some degree by the corresponding oxazol-5(4H)-ones (azlactones). Similar characteristics would be anticipated in the corresponding oxazol-4(5H)-ones, and we now describe their ready cycloaddition to acetylenic dipolarophiles to give substituted furans.

2-Phenyloxazol-4(5H)-one³ (1) when refluxed in acetic anhydride (or xylene) with a slight excess of dimethyl acetylenedicarboxylate for 12 h gave, after chromatography, dimethyl 2-phenylfuran-3,4-dicarboxylate (5; $R = CO_2Me$) (50% yield) m.p. 70–71°. Similarly dibenzoylacetylene gave 3,4-dibenzoyl-2-phenylfuran (5; R = Bz) (30%) m.p. 148·5–149°. The products were identified by their i.r., n.m.r., and mass spectra.

In contrast to the oxazol-5(4H)-ones, two reaction pathways are feasible with the present system. Tautomerism between the oxazol-4(5H)-one (1) and the oxazolium-4-oxide (1a), for which the resonance formula (1b) indicates contributions from a heteroaromatic carbonyl ylide, accounts for a 1,3-dipolar type cycloaddition via an intermediate

such as (2). Alternatively, the enolic tautomer⁴ (3) may be involved in a Diels-Alder type addition to an intermediate (4) which, by loss of HOCN, would give the furan (5). The latter pathway cannot be discounted, as substituted 5-ethoxyoxazoles^{2,5} undergo Diels-Alder additions with dienophiles.

We thank the U.S. Public Health Service, National Cancer Institute, for financial support; J. M. acknowledges the award of an NDEA Traineeship.

(Received, 12th July 1972; Com. 1207.)

² H. Gotthardt, R. Huisgen, and H. O. Bayer, J. Amer. Chem. Soc., 1970, 92, 4340.

Y. S. Rao and R. Fuller, Chem. Comm., 1970, 1622.
M. J. Nye and W. P. Tang, Tetrahedron, 1972, 28, 455.

¹ R. Huisgen, E. Funke, H. Gotthardt, and H.-L. Pauke, *Chem. Ber.*, 1971, 104, 1532; E. Funke, R. Huisgen, and F. C. Schaefer *ibid.*, p. 1550; E. Brunn, E. Funke, H. Gotthardt, and R. Huisgen, *ibid.*, p. 562; H. O. Bayer, R. Huisgen, R. Knorr, and F. C. Schaefer, *ibid.*, 1970, 103, 2581.

^b R. Grigg and J. L. Jackson, J. Chem. Soc. (C), 1970, 552; G. Ya. Kondrat'eva and C. H. Huang, Doklady Akad. Nauk S.S.S.R., 1962, 142, 593.